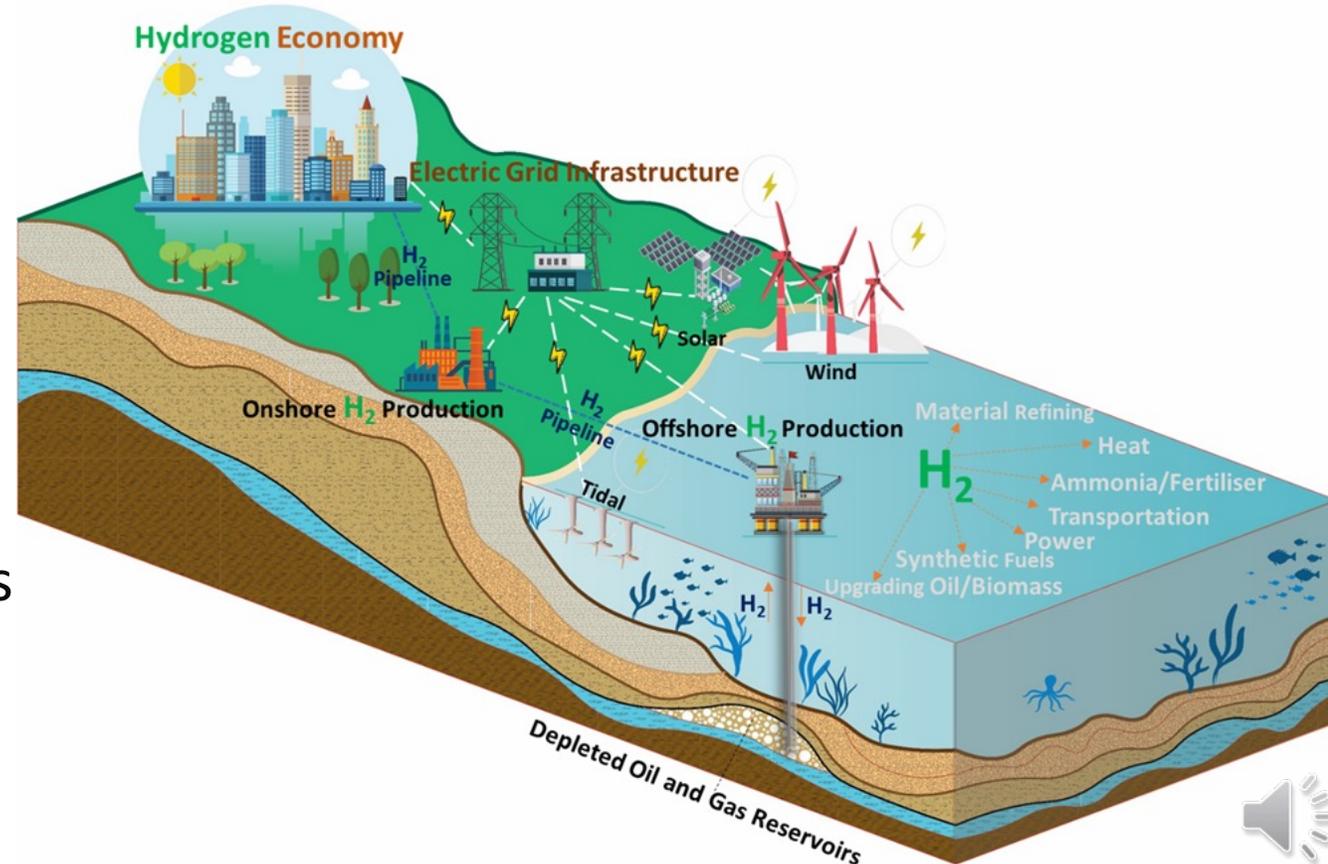
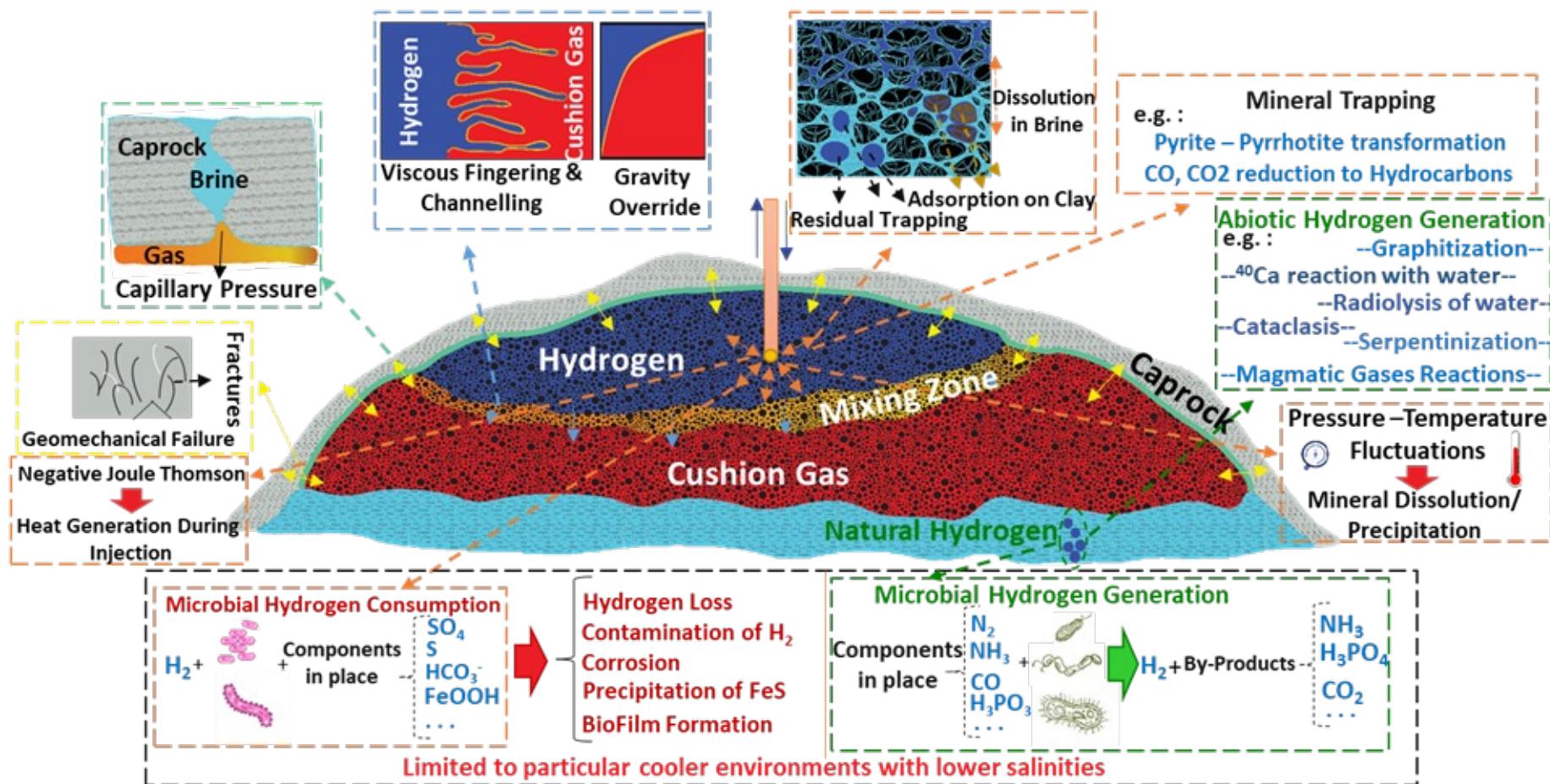


# Implications of Hydrogen's Physical and Chemical Reactivity for Geological Storage

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School of GeoSciences  
University of Edinburgh

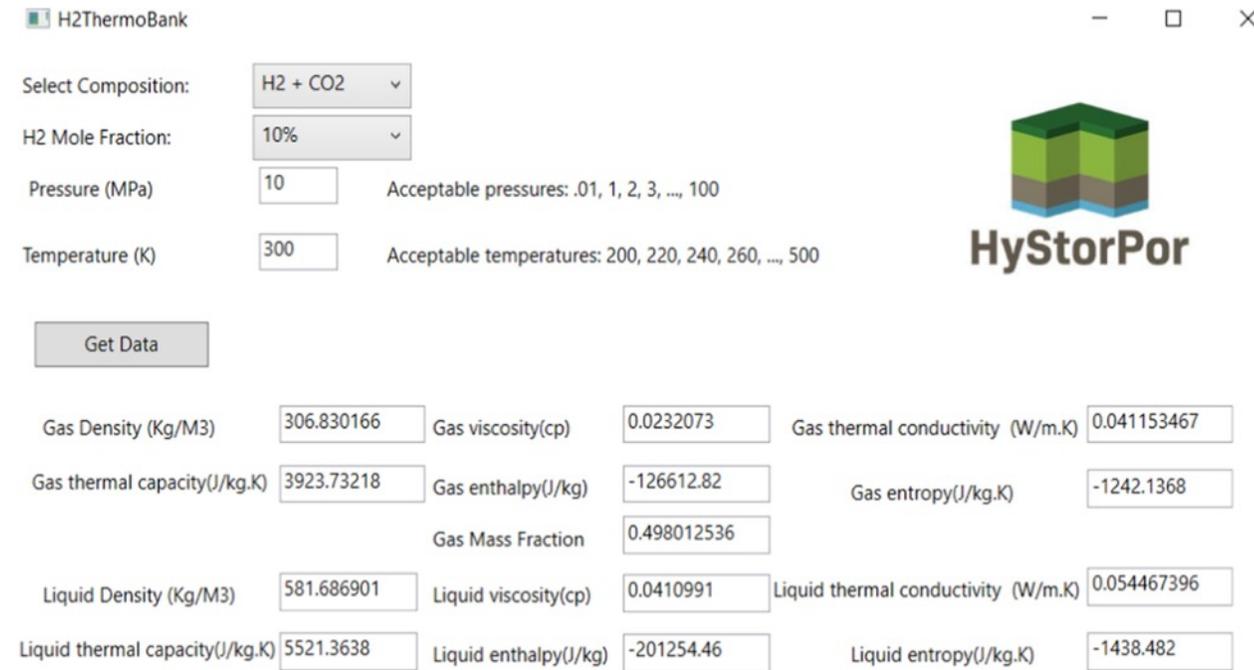


## What Processes Will Impact Geological H<sub>2</sub> Storage?



## H2Thermobank: Thermodynamics of Hydrogen Gas Streams

- Thermodynamic properties of hydrogen mixtures, including CO<sub>2</sub>, N<sub>2</sub>, CH<sub>4</sub> and a typical UK North Sea Natural Gas.
- Mole fractions of hydrogen from 10-90 mole %.
- Pressure from 0.01 – 100 Mpa.
- Temperatures from 200-500 K (-73C to 227C).



**H2ThermoBank**

Select Composition:

H2 Mole Fraction:

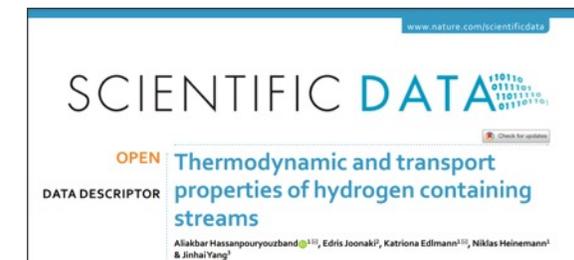
Pressure (MPa)  Acceptable pressures: .01, 1, 2, 3, ..., 100

Temperature (K)  Acceptable temperatures: 200, 220, 240, 260, ..., 500

Gas Density (Kg/M3)	<input type="text" value="306.830166"/>	Gas viscosity(cp)	<input type="text" value="0.0232073"/>	Gas thermal conductivity (W/m.K)	<input type="text" value="0.041153467"/>
Gas thermal capacity(J/kg.K)	<input type="text" value="3923.73218"/>	Gas enthalpy(J/kg)	<input type="text" value="-126612.82"/>	Gas entropy(J/kg.K)	<input type="text" value="-1242.1368"/>
		Gas Mass Fraction	<input type="text" value="0.498012536"/>		
Liquid Density (Kg/M3)	<input type="text" value="581.686901"/>	Liquid viscosity(cp)	<input type="text" value="0.0410991"/>	Liquid thermal conductivity (W/m.K)	<input type="text" value="0.054467396"/>
Liquid thermal capacity(J/kg.K)	<input type="text" value="5521.3638"/>	Liquid enthalpy(J/kg)	<input type="text" value="-201254.46"/>	Liquid entropy(J/kg.K)	<input type="text" value="-1438.482"/>



Hassanpouryouzband et al. "Thermodynamic and transport properties of hydrogen containing streams." Scientific Data 7.1 (2020): 1-14. doi.org/10.1038/s41597-020-0568-6  
<https://github.com/aliakbarhssnpr/H2ThermoBank>

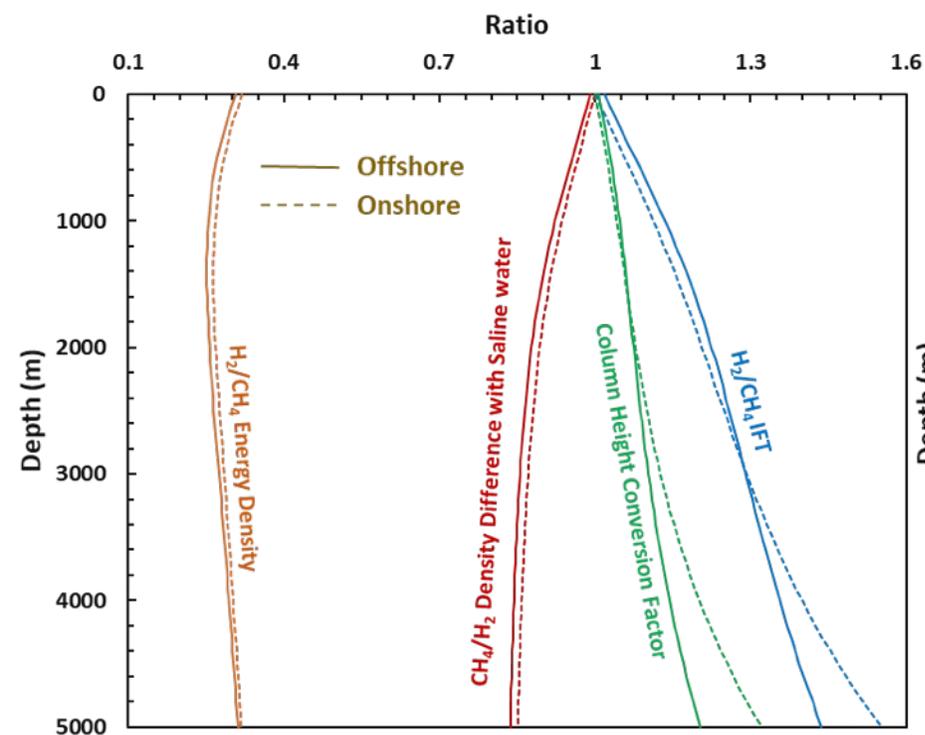


## Hydrogen Caprock Sealing

- Column heights reflect the sealing capacity of any caprock.
- Column height conversion factor calculated to convert known natural gas column heights to hydrogen column heights.
- Hydrogen can be stored at a higher pressure in the reservoir than methane.

$$\Psi_{CH_4/H_2} = \frac{\Delta\rho_{CH_4/water} \gamma_{H_2/water} \cos\theta_{H_2/water}}{\Delta\rho_{H_2/water} \gamma_{CH_4/water} \cos\theta_{CH_4/water}}$$

*Density difference*       $\frac{IF}{T}$       *Wettability contact angle*



# Storage Capacity Estimation

H2CapeEs

H2CapEs

HyStorPor

Select Storage Type

Pipe Length(Km)  Pipe Diameter (m)

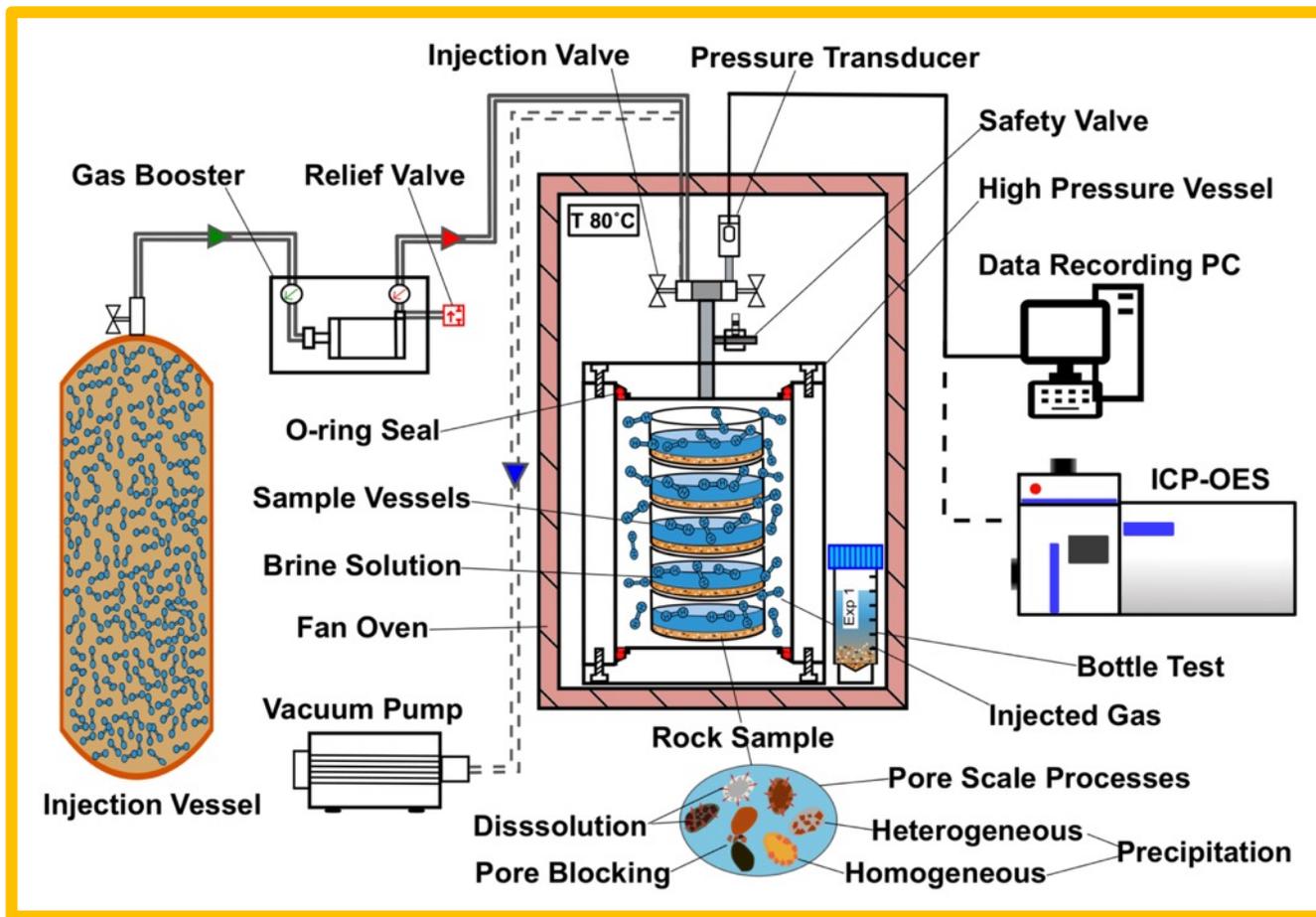
Pressure (Mpa)  Temperature (K)

**Get Energy Storage Capacity**

Energy (TWH):



## Geo-Chemistry Experiments



Over 500 experiments

**8 Different Sandstones**

Temperatures: 50-80 ° C

Pressures: 0.01-20 Mpa

Salinities: 0-25 weight% NaCl

Different Rock Sizes

Different Water/Rock Ratios

...

**12 Different Samples from gas fields**

Experiments at Reservoir Condition

**4 Caprock Sample**

**3 Cement Samples**

**Pure Minerals**

**Gypsum, Calcite, Dolomite, and Pyrite**



# Geo-Chemistry Experiments



## Temperature

- Experiments undertaken in an thermostatically controlled oven where temperatures up to 80°C are reliable and repeatable due to precise temperature regulation and monitoring which limited any potential temperature-dependent effects on potential geochemical reactions.



## Pressure

- Bottle tests undertaken in sealed atmospheric plastic containers with no injected gas accounted for any pressure-dependent effects on mineral reactions in each experiment. Therefore, experiments at pressures between 1 - 20 MPa were validated and deemed repeatable.



## Salinity

- Experiments at salinities up to 250 ppt are reliable and repeatable.



## Sample sterilization

- No effect on the results were observed due to sample sterilisation. Sterilisation is an essential preparation step for the rock samples to minimise the potential occurrence of biotic reactions that may influence component concentrations, particularly the gas composition.



## Rock to water ratio

- Each sample was examined at different rock-water ratios to evaluate the rate-dependent effect of mineral phase concentration on hydrogen associated geochemical reactions.



## Disaggregated grain size

- Grain sizes from 0.335 to 4 mm are suitable however, larger grain sizes must be balanced by a higher rock to water ratio. Smaller grain sizes increase mineral reactive surface areas resulting in higher concentrations of dissolved components. Uniform grain size across all experimental conditions for an individual rock type was implemented to ensure a more robust analysis of sample reactivity.



## Reaction vessel

- High-pressure/temperature, 316 stainless steel, static batch reactor with O-ring seals and high tensile cap screws. No degradation or blistering was observed on the steel or O-rings under this range of parameters.



## Sample vessel

- Tested with steel - contamination identified - changed to glass bottles, self-standing plastic centrifuge bottles used for atmospheric tests



## Oxygen removal

- Nitrogen flow through and vacuum pumping was utilised on all experiments to remove free oxygen from the vessels before gas injection, controlling the redox-sensitive nature of anoxic subsurface environments.



## Control experiments

- All samples were run with hydrogen and nitrogen at pressure and temp and with just brine at experimental temperature but ambient pressure to ensure any reactions observed could be directly attributable to hydrogen rather than say pressure or brine.



## NaCl/brine quality

- The sensitivity of our equipment is such that it will pick up on even the smallest variations in the composition of the NaCl that we used to make the brine..

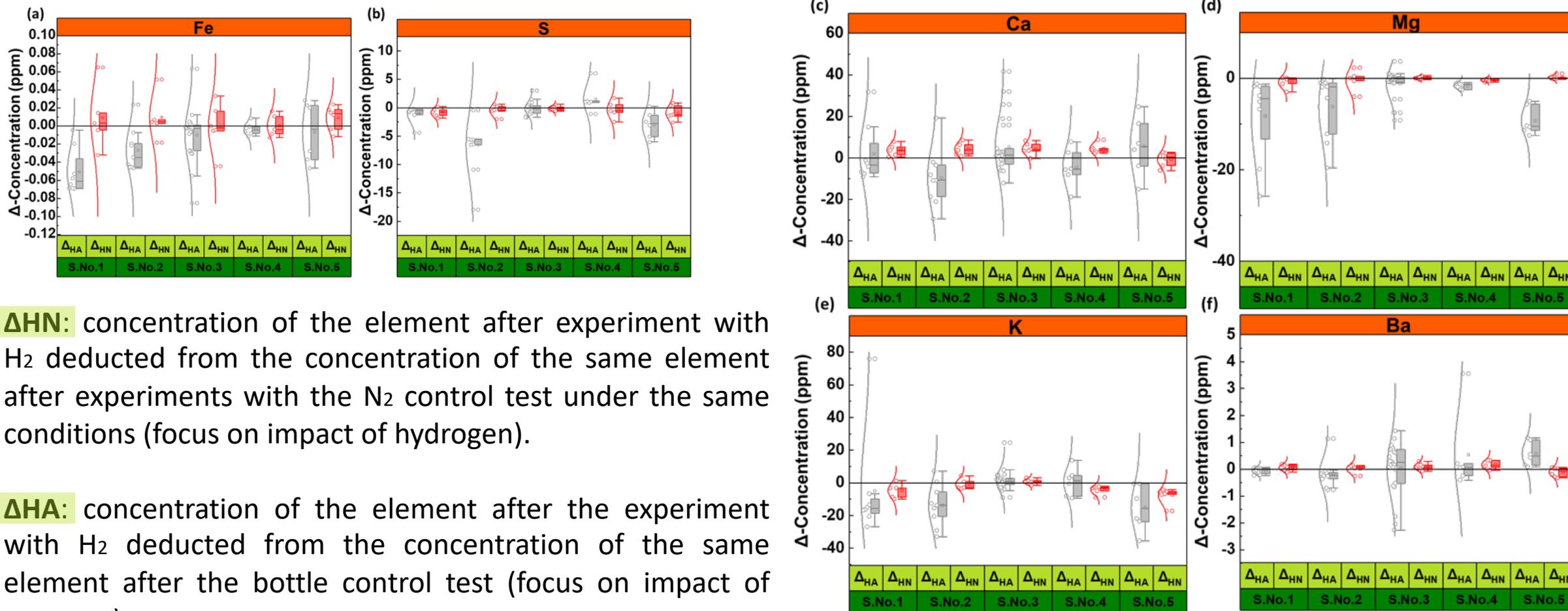


## Repeatability

- Overall experimental repeatability is high with robust results produced across 253 experiments including 40 repeats covering all conditions. Replacement of batch reactor O-ring seals after each experimental cycle.



## Results: Element Fluid Concentrations On Exposure To Hydrogen (From Icp-oes)

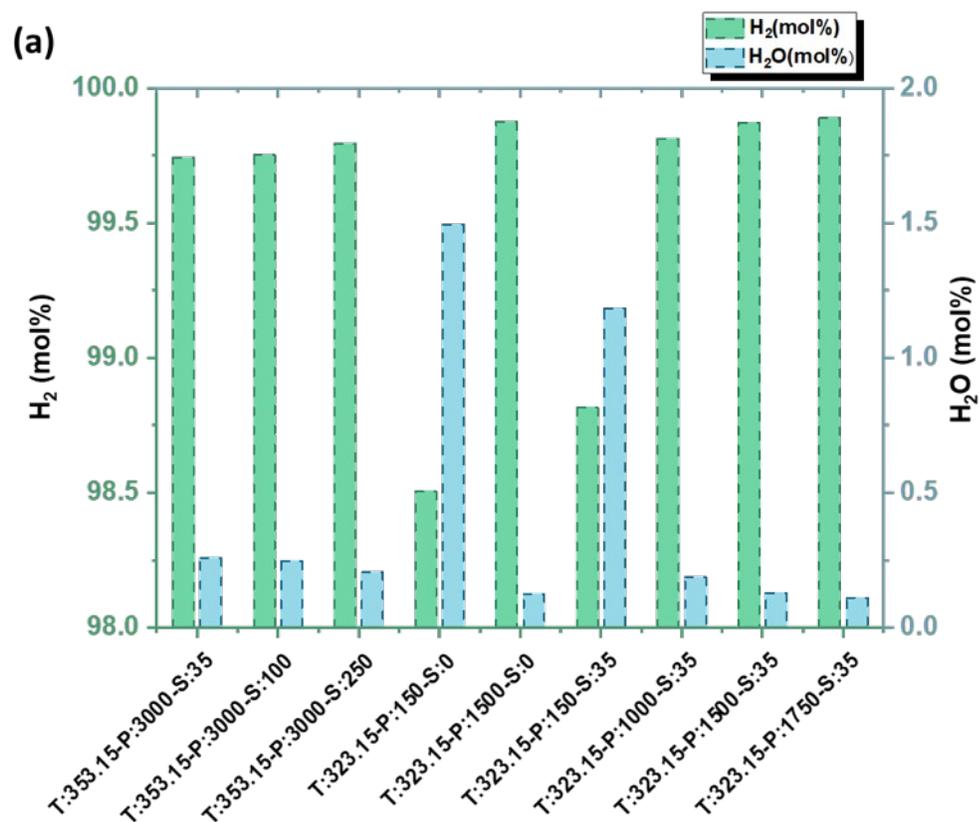


**$\Delta_{HN}$ :** concentration of the element after experiment with H<sub>2</sub> deducted from the concentration of the same element after experiments with the N<sub>2</sub> control test under the same conditions (focus on impact of hydrogen).

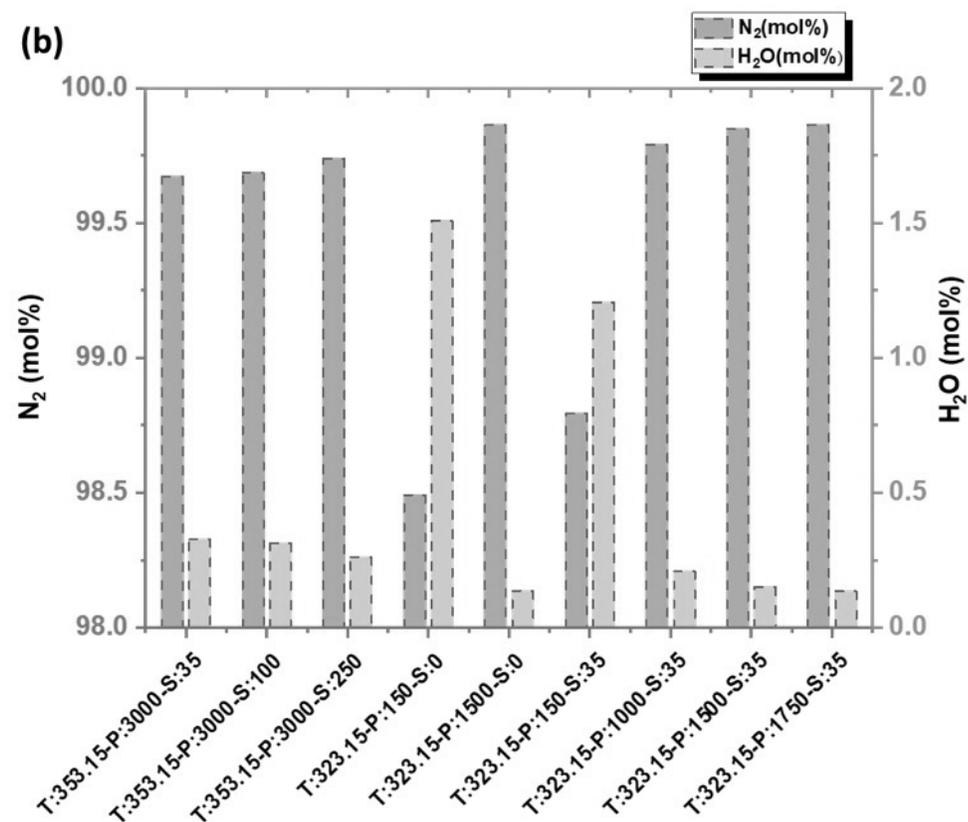
**$\Delta_{HA}$ :** concentration of the element after the experiment with H<sub>2</sub> deducted from the concentration of the same element after the bottle control test (focus on impact of pressure).



## Results: Gas Composition



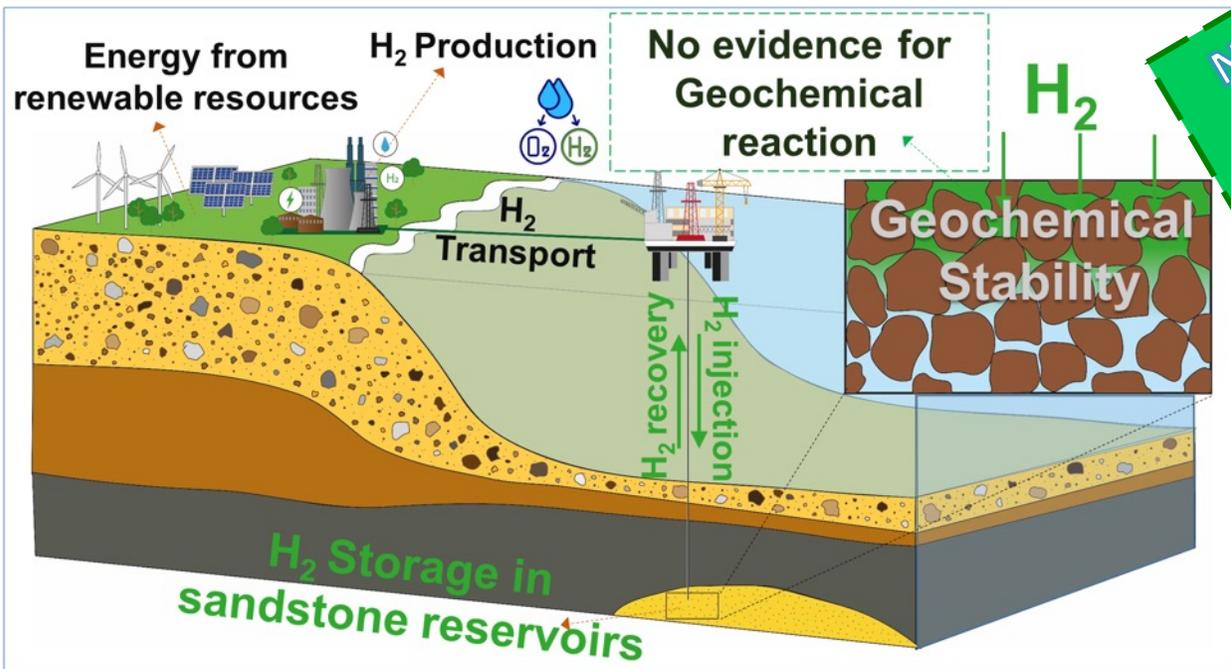
Hydrogen gas purity experiments



Nitrogen gas purity experiments



## Geo-Chemistry Experiments



No evidence of significant geochemical reactions so far ...

Hassanpouryouzband et al. "Geological hydrogen storage: geochemical reactivity of hydrogen with sandstone reservoirs" ACS Energy Letters, 2022, doi.org/10.1021/acsendergylett.2c01024

ACS Energy LETTERS

Geological Hydrogen Storage: Geochemical Reactivity of Hydrogen with Sandstone Reservoirs

Aliakbar Hassanpouryouzband,\* Kate Adie, Trystan Cowen, Eike M. Thaysen, Niklas Heinemann, Ian B. Butler, Mark Wilkinson, and Katriona Edlmann\*

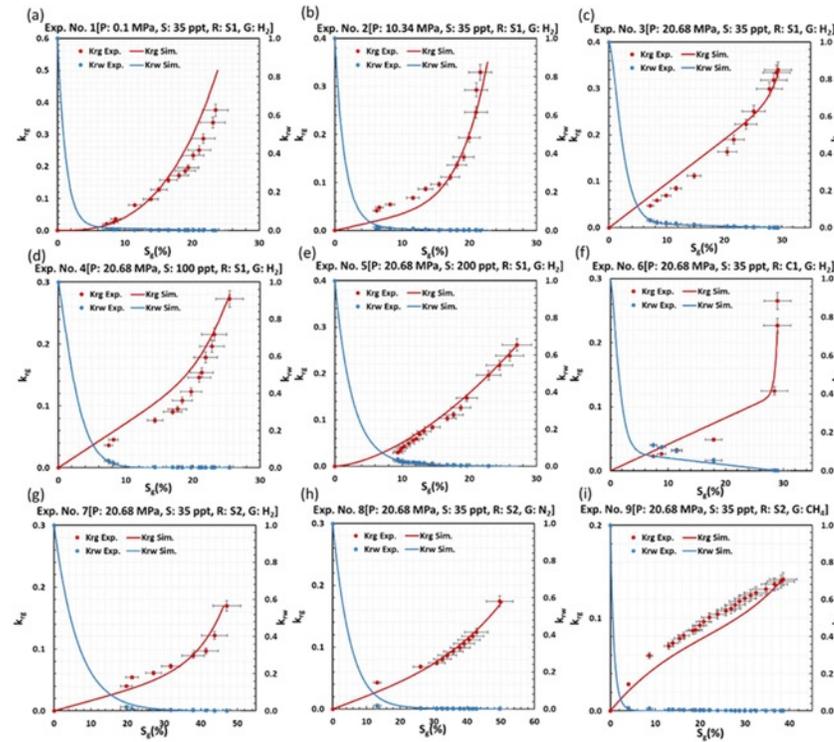
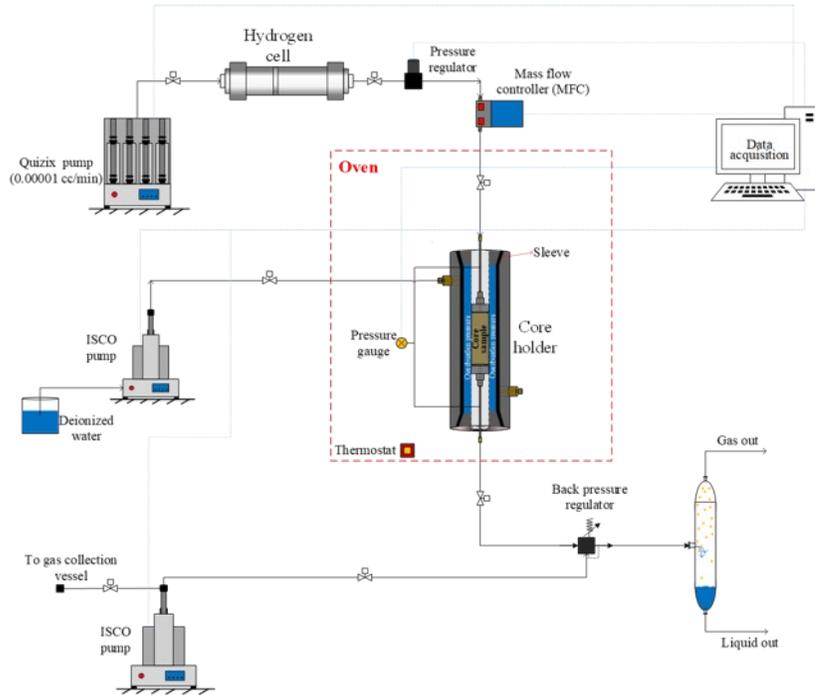
ENVIRONMENTAL LETTERS

Geochemical Integrity of Wellbore Cements during Geological Hydrogen Storage

Adnan Aftab, Aliakbar Hassanpouryouzband,\* Abby Martin, Jackie E. Kendrick, Eike M. Thaysen, Niklas Heinemann, James Utley, Mark Wilkinson, R. Stuart Haszeldine, and Katriona Edlmann\*



## Unsteady State Relative Permeability Experiments



2 Different Sandstone Cores from Gas Reservoir

Temperatures: 80 ° C

Pressures: 0.01, 10, 20 Mpa

Salinities: 3.5, 10, 20 weight% NaCl

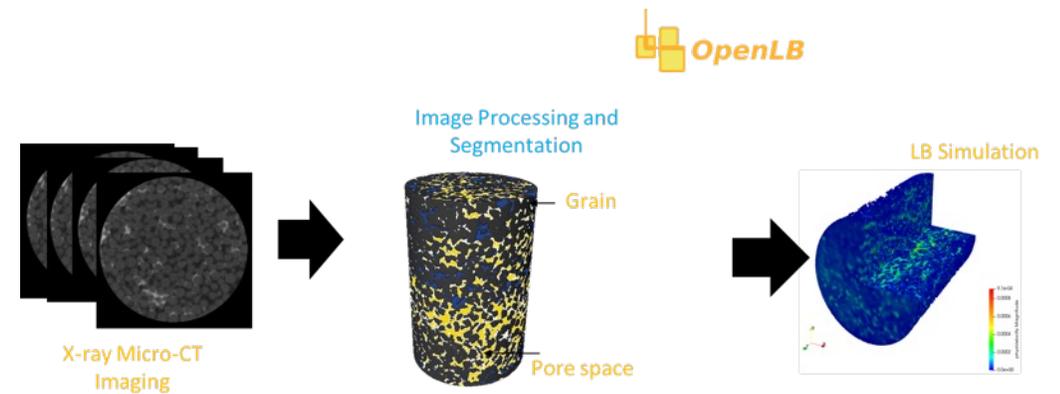
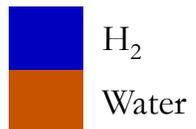
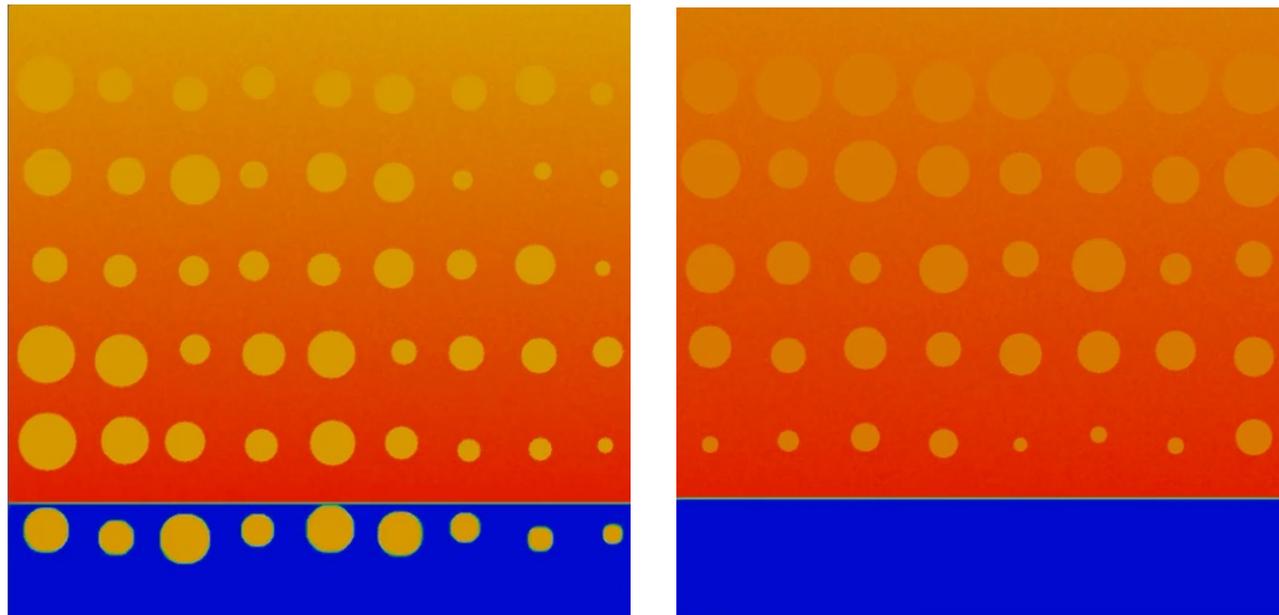
Using Different Gas for Comparison: N2, CH4

Carbonate Rock from Gas Reservoir

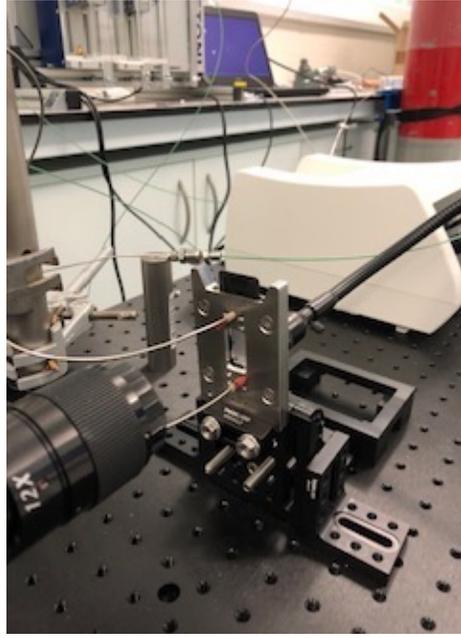
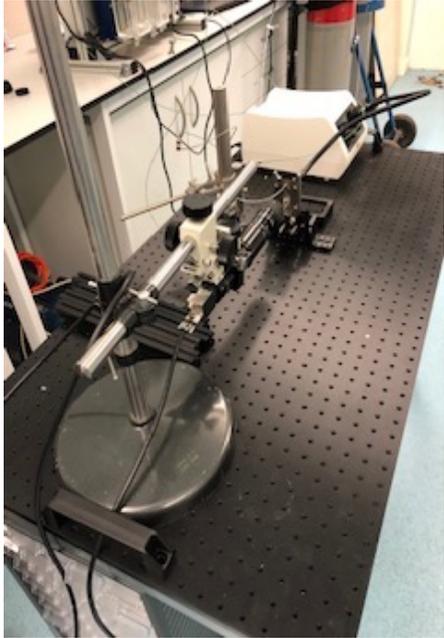
MICP, XRD, Centrifuge g-w drainage



# Pore-scale Simulation of Hydrogen Flow: The Effect Of The Grain Arrangement

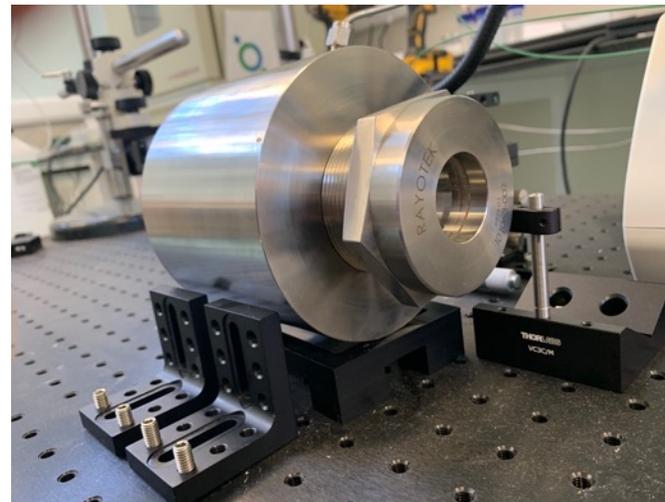


## Glass Micromodel Hydrogen Flow Experiments



- Pressure range in existing cell up to 100 bar
- Multiphase flow
- Flow image recording
- Micromodels etched for any pore network
- Micromodels can have different wettability

## High pressure and temperature visual cells



- Pressure range in existing cell up to 270 bar (27 MPa)
- Wettability measurements
- Image recording



# Thank you for listening

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